# **ALKALOIDS FROM MICROPLUMERIA ANOMALA\***

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**Key Word Index**—Microplumeria anomala; Apocynaceae; indoline alkaloids; anomaline; demethoxyanomaline; 12-O-methylanomaline; NMR and mass spectra.

Abstract—Three indoline alkaloids have been isolated from the Amazonian plant Microplumeria anomala and their structures determined by spectral means.

## INTRODUCTION

In continuation of a study of alkaloid constituents of the Apocynaceae an investigation of the tree *Microplumeria anomala*, indigenous to the Amazon region, was undertaken. The present communication reports the isolation and characterization of three bases from this plant.

### RESULTS AND DISCUSSION

Extraction of the bark with ethanol and chromatography of the extract on alumina and silica gel yielded a  $C_{22}H_{30}O_4N_2$ , crystalline compound mp 216-218°. Its IR spectrum (KBr) exhibited a broad OH absorption band (3400 cm<sup>-1</sup>) and strong amide carbonyl absorption (1630, 1600 and 1580 cm<sup>-1</sup>) and its UV spectrum (methanol) showed maxima at 226 nm ( $\log \varepsilon 4.27$ ) and 259 nm ( $\log \varepsilon 3.85$ ). The <sup>1</sup>H NMR spectrum of a deuterochloroform solution of the alkaloid indicated the presence of a methoxyl singlet ( $\delta$  3.88), an acetyl methyl singlet ( $\delta$  2.33), a C-ethyl group bonded to a non-protonated carbon [0.66 (t, 3H, J = 7 Hz, Me), 1.37] $(dq, 2H, J = 16, 7 Hz, CH_2)$  and two vicinal aromatic hydrogens [6.62 (d, 1H, J = 8 Hz), 6.69 (d, 1H, J= 8 Hz)]. These data were reminiscent of the spectral characteristics of the Aspidosperma indoline alkaloid aspidocarpine (1a) [1, 2] and suggested the new natural base, named herewith anomaline, to be a hydroxyaspidocarpine. For this reason an exhaustive <sup>1</sup>H NMR spectral analysis of aspidocarpine (1a) was undertaken.

The <sup>1</sup>H NMR spectrum of a deuterochloroform solution of aspidocarpine (1a) showed signals at 0.64 (t, 3H, J = 7 Hz, ethyl Me), 1.38 (q, 2H, J = 7 Hz, ethyl CH<sub>2</sub>), 2.23 (s, 1H, H-21), 2.32 (s, 3H, Ac Me), 3.05 (ddd, 1H, J = 10, 3, 2 Hz, H-3 or H-5), 3.12 (dt, 1H, J = 9, 3 Hz, H-3 or H-5), 3.88 (s, 3H, OMe), 4.06 (dd, 1H, J = 11, 6 Hz, H-2), 6.61, 6.70 (d, 1H each, J = 8 Hz, aromatic Hs) and 8.91 (s, 1H, OH), whose similarity with the hydrogen shifts of anomaline (1b) (see above) [2.49 (m, 1H, H-3 or H-5), 2.62

(s, 1H, H-21), 2.84 (ddd, 1H, J = 11, 3, 2 Hz, H-3 or H-5), 3.11 (dt, 1H, J = 9, 3 Hz, H-3 or H-5), 3.70 (br s, 1H, H-15), 4.10 (dd, 1H, J = 11, 6 Hz, H-2) showed the latter to be indeed a hydroxyaspidocarpine. The addition of pyridine caused deshielding of the methylene unit of the quaternary ethyl group by 0.12 ppm [3] indicating the proximity of the latter to the non-aromatic hydroxyl group. Anomaline's aspidocarpine-like structure was confirmed by its mass spectrum. By analogy with the behavior of all Aspidosperma alkaloids the spectrum showed five characteristic fragments, the [M]<sup>+</sup> and cations 2–5 (R<sub>1</sub> = H, R<sub>2</sub> = OMe) at m/z 386 (97%), 358 (11), 190 (3), 176 (4) and 140 (100), respectively. These results corroborated the aspidocarpine-like structure and restricted the nonaromatic hydroxyl group to the piperidine ring in the vicinity of the angular ethyl group. The specific location of the alcohol function was derived from <sup>13</sup>C NMR data (see below).

The anomaline-bearing, Amazonian plant yielded two, minor cogeners, demethoxyanomaline (1c) and 12-0methylanomaline (1d) from the leaves, whose structural relationship with the major bark alkaloid was readily apparent from their spectral characteristics. Thus, the first base lacked a methoxyl signal in its <sup>1</sup>H NMR spectrum and showed an altered aromatic hydrogen region in the spectrum [6.70 (d, 1H, J = 8 Hz, H-9), 6.82 (d, 1H, J= 8 Hz, H-11) and 7.06 (t, 1H, J = 8 Hz, H-10) ], while thesecond base revealed a two-methoxyl singlet ( $\delta$  3.88) and two vicinal, aromatic hydrogen doublets ( $\delta$  6.65 and 6.85, J = 8 Hz each). The IR spectra were anomaline-like. The UV spectrum (methanol) of 1c exhibited maxima at 221, 260, 288 nm (log  $\varepsilon$  4.13, 3.72, 3.47, respectively) and that of **1d** at 221, 253, 288 nm (log  $\varepsilon$  4.29, 3.80, 3.49, respectively). These facts showed the substances to differ from anomaline in their aromatic substituents, as denoted in their names. The mass spectra of the minor bases confirmed these differences even more clearly. Thus although the compounds showed base peaks identical with anomaline (m/z 140); fragment 5), the [M]<sup>+</sup> and fragments 2-4 of demethoxyanomaline  $[m/z^{-3}5\overline{6} (34\%), 3\overline{2}8 (8), 160 (4),$ 146 (3) differed from those of anomaline by -30 mu and the  $[M]^+$  and ions 2-4 of 12-0-methylanomaline [m/z] 400 (49%), 372 (17), 204 (5), 190 (4)] varied from those of the major alkaloid by +14 mu.

With the gross structures of the three alkaloids es-

<sup>\*</sup>Part 78 in the series "Carbon-13 Nuclear Magnetic Resonance Spectroscopy of Naturally Occurring Substances". For part 77 see McPhail, A. T., Hagaman, E. W., Kunesch, N., Wenkert, E. and Poisson, J. (1983) *Tetrahedron* 39 (in press).

$$\begin{array}{c} R_{2} \xrightarrow{11} \\ R_{1}O \\ O = \\ \end{array} \xrightarrow{\left\{\begin{matrix} 10 \\ 13 \end{matrix}\right\}} \left\{\begin{matrix} 6 \\ 121 \\ 13 \end{matrix}\right\} \left\{\begin{matrix} 120 \\ 14 \end{matrix}\right\}} \\ R_{3} \\ \end{array}$$

**1a** 
$$R_1 = R_3 = H, R_2 = OMe$$

**1b** 
$$R_1 = H, R_2 = OMe, R_3 = OH$$

$$1c R_1 = R_2 = H, R_3 = OH$$

**1d** 
$$R_1 = Me$$
,  $R_2 = OMe$ ,  $R_3 = OH$ 

tablished, it remained for  $^{13}\text{C}$  NMR spectroscopy to remove any structural uncertainties. In order to assign the aromatic carbon shifts, it was necessary to rely on shift data for  $N_a$ -methyl- $2\beta$ ,16 $\beta$ -dihydrotabersonine (6) [4], a retuline derivative (7) [5] and a vobtusine derivative (8) [6], wherefrom it was possible to make complete shift assignments for the heretofore uninspected indoline alkaloids refractine (9) [7] and cylindrocarpine (10) [2, 8]. The non-aromatic carbon shifts of 9 were based on those of venalstonine [9] and those of 10 on the shift data of a variety of Aspidosperma bases [10].

The aromatic carbon shifts of 12-0-methylanomaline (1d) could be assigned with the use of 9 as a model and by the application of methoxyl substituent parameters [11] and these are listed on formula 11. The  $\delta$  values for the aromatic carbons of anomaline (1b) (methines: 109.9, 112.8; non-protonated carbons: 127.4, 132.8, 137.1, 149.2) and demethoxyanomaline (1c) (methines: 113.6, 117.5, 127.8; non-protonated carbons: 126.8, 140.6, 147.0) could not be assigned with certainty and will require a future model study for proper designation. The identity of the aromatic carbon shifts of anomaline (1b) with those of aspidocarpine (1a) revealed the nature of the aromatic substitution pattern of the former base.

The non-aromatic carbon shifts of aspidocarpine (1a), anomaline (1b) and its two cogeners 1c and 1d, listed in Table 1, were based on shift data for various Aspidosperma alkaloids [10] and revealed the position of the hydroxyl group of the three new bases as C-15.

Furthermore, a comparison of the  $\delta$  values of the three new compounds (1b-d) with those of aspidocarpine (1a) showed the 15-hydroxyl group to be oriented axially

Table 1. Non-aromatic carbon shifts of alkaloids

1a-1d\*

Carbon	1a	1b	1e	1d
2	70.0†	70.1	69.5	69.1
3	53.4	47.0	47.0	46.9
5	52.1	51.8	51.9	51.6
6	39.1	39.1	38.9	37.9
7	51.9	51.8	52.5	51.6
14	24.8	29.2	29.2	29.0
15	33.7	68.0	68.0	68.1
16	21.2	21.8	21.8	21.7
17	22.6	22.8	22.8	22.8
18	6.5	6.0	6.0	6.1
19	29.7	24.7	24.7	24.5
20	35.2	40.2	40.2	40.2
21	70.3†	65.8	65.9	66.2

\*In CDCl<sub>3</sub> solutions; chemical shifts in ppm downfield from TMS;  $\delta$  (TMS) = (CDCl<sub>3</sub>) + 76.9 ppm. The acetyl C=O and Me shifts of 1a-c are 169.1 and 22.5 ppm, respectively. The OMe shifts of 1a-b are 56.1 ppm.

†The signals may be interchanged.

within the conformational framework 12. Such conformation accounted for the shielding (i.e. y shifts) of C-3, C-19 and C-21 on introduction of the hydroxyl group into the piperidine ring. The combined data, and especially the <sup>13</sup>C NMR analyses, prove the three new Aspidosperma bases to possess the relative configurations depicted in formulas 1b-d.

## **EXPERIMENTAL**

MPs are uncorr. IR spectra were recorded in KBr pellets and UV spectra in MeOH solns. <sup>1</sup>H NMR spectra of CDCl<sub>3</sub> solns with TMS as int. standard ( $\delta = 0$  ppm) were measured at 360 MHz. <sup>13</sup>C NMR spectra were obtained on a wide-bore, broad-band spectrometer, operating at 50.31 MHz in the Fourier transform mode. The carbon shifts denoted on formulas 6–11 are from CDCl<sub>3</sub> solns in ppm downfield from TMS;  $\delta$  (TMS) =  $\delta$  (CDCl<sub>3</sub>) + 76.9 ppm. The asterisked  $\delta$  values of formula 9 may be interchanged.

Isolation of anomaline (1b). A M. anomala (M. Arg.) Mgf. plant (common name: cururu) (INPA-Manaus herbarium no. 78.959) was collected along the bank of the Rio Negro in the vicinity of Manaus. Its ground, dried bark (3.5 kg) was extracted twice for 8 days with 8 l. of EtOH. The extract was concd to one-third its vol. and filtered. Chromatography of the ppt (25 g) showed it to be a mixture of lupeol, lupenone and sitosterols. Vacuum evaporation of the filtrate yielded 165 g (5.5% dry bark) of residue. Chromatography of a portion (10 g) of the latter on neutral  $Al_2O_3$  and elution with  $C_6H_6$  gave more of the steroids and

triterpenes, whereas elution with CHCl<sub>3</sub> yielded alkaloidal material. Prep. TLC of the latter on silica gel and development with CHCl<sub>3</sub>-MeOH (9:1) yielded a solid, which on crystallization from EtOAc gave colorless needles of anomaline (1b), mp 216-218°. Found (exact mass): m/z 386.22053. C<sub>22</sub>H<sub>30</sub>N<sub>2</sub>O<sub>4</sub> requires: m/z 386.22053.

Isolation of demethoxyanomaline (1c) and 12-O-methylanomaline (1d). Dried, powdered leaves (680 g) were sprayed with conc Na<sub>2</sub>CO<sub>3</sub> soln (100 ml), left for 18 hr and then extracted twice for one week at room temp. with 10 l. of Et<sub>2</sub>O. The extract was extracted exhaustively with 5 % HCl soln. After adjustment of the aq. extract to pH 10 with NH4OH it was extracted thoroughly with CHCl<sub>3</sub>. The extract was evaporated and the residue (10.6 g, 1.6 % of dry leaves) chromatographed on neutral Al<sub>2</sub>O<sub>3</sub>. Elution with C<sub>6</sub>H<sub>6</sub> gave steroids and triterpenes, while elution with CHCl<sub>3</sub> yielded an alkaloid mixture. Prep. TLC of the latter on silica gel and development with CHCl<sub>3</sub>-MeOH (9:1) led to an alkaloid-rich fraction, which was chromatographed on silica gel. Elution with C<sub>6</sub>H<sub>6</sub> removed more non-polar, nonalkaloidal material, while elution with CHCl<sub>3</sub>-C<sub>6</sub>H<sub>6</sub> (4:1), CHCl<sub>3</sub> and CHCl<sub>3</sub>-MeOH (9:1) yielded alkaloid fractions, whose constituents were monitored by <sup>1</sup>H NMR spectroscopy, especially the OMe and OH signals.

Crystallization of the constituents of the early eluates with EtOAc gave colorless needles of demethoxyanomaline (1c), mp 263-265°. IR  $\nu$  (cm<sup>-1</sup>) 3520 (OH), 1620, 1600, 1580 (C=O). <sup>1</sup>H NMR  $\delta$  0.70 (t, 3H, J = 7 Hz, Me), 2.40 (s, 3H, Ac Me), 2.52, 2.85, 3.14 (m, 1 each, CH<sub>2</sub>NCH<sub>2</sub>), 2.68 (s, 1H, H-21), 3.70 (s, 1H, OCH), 4.13 (m, 1H, H-2). Found (exact mass): m/z 356.2100.

 $C_{21}H_{28}N_2O_3$  requires: m/z 356.2100. Crystallization of the constituents of the middle eluates with EtOAc yielded colorless needles of anomaline (1b), mp 216–218° (see above), while the later eluates with the same solvent led to the isolation of amorphous 12-O-methylanomaline (1d). IR v (cm<sup>-1</sup>) 3430 (OH), 1630 (br, C=O); <sup>1</sup>H NMR  $\delta$  0.66 (t, 3H, J = 7 Hz, Me), 2.25 (s, 3H, acetyl Me), 2.48, 2.82, 3.11 (m, 1H each, CH<sub>2</sub> NCH<sub>2</sub>), 2.59 (s, 1H, H-21), 3.67 (s, 1H, OCH), 3.80 (m, 1H, H-2). Found (exact mass): m/z 400.2383.  $C_{23}H_{32}N_2O_4$  requires: m/z 400.2362.

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